# Joalin, the First Nitrogen-containing Xenicane Diterpene isolated from a Brown Seaweed collected off the Senegalese Coast 

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Joalin (-)-1 isolated from a brown seaweed, Dictyota sp., collected off the Senegalese coast, is a rare xenicane-derived diterpene in that it contains nitrogen.

Brown seaweeds of the order Dictyotales, sea hares and alcyonaceans, have given a plethora of xenicane diterpenes containing no nitrogen, ${ }^{1}$ a feature also of other cembranederived diterpenes, ${ }^{1}$ the sole exception being a family of aminocembranolides recently discovered in a tropical alcyonacean. ${ }^{2}$ Sharp contrast is thus apparent with the nitrogen-containing diterpenes of sponges belonging to the orders Axinellida ${ }^{3}$ and Haplosclerida. ${ }^{4}$ It is interesting, therefore, to have now found in a seaweed, collected off the Senegalese coast, the first nitrogen-bearing xenicane diterpene, joalin (-)-1, which is reported here.

The MeOH extract of the seaweed, Dictyota sp. (Dictyotales), was subjected to extensive chromatography to give joalin $(-)-1$, whose composition $\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{NO}_{3}$ was deduced from mass and ${ }^{13} \mathrm{C}$ NMR spectroscopy. Together with ID (assignments from differential double irradiations) and 2D (COSY 120) ${ }^{51} \mathrm{H}$ NMR spectra, this showed that three of the seven unsaturated positions of joalin are accounted for by a tetra substituted and two tri substituted $\mathrm{C}=\mathrm{C}$ bonds. An ester or amide/lactam carbonyl group was also suggested by the signal $\delta_{\mathrm{C}} 174.60$ (s). Choice of the latter was dictated by IR absorption at $v_{\max } 1702 \mathrm{~s}$ $\mathrm{cm}^{-1}$, and the presence of one nitrogen atom was confirmed by HREIMS data. Therefore, joalin must be tricyclic. The portion from $\mathrm{C}-1$ to $\mathrm{C}-4$ and bonding of $\mathrm{C}-9$ to $\mathrm{C}-1$, was based on NMR data for the deshielded protons $4-\mathrm{H}, 9-\mathrm{H}$ and $18-\mathrm{H}$ whose homo- and hetero-nuclear NMR correlations (HMQC and $\mathrm{HMBC})^{6}$ are indicated on the structural formula by double arrows. Thus, oxygen substitution at $\mathrm{C}-4$ was suggested by lowfield resonances for both this carbon and 4-H. Moreover,

$(-)-1$


1 a

the signals $\delta_{\mathrm{C}} 79.47$ (d) and $\delta_{\mathrm{H}} 4.52$ (td) suggested $\mathrm{C}-9$ to be an allylic carbon bearing a methoxy group long-range heterocoupled with $9-\mathrm{H}$. Finally, C-18 bound to both N and O was suggested by lowfield resonance for both this carbon and $18-\mathrm{H}$. Further support for the connectivity was afforded by W H(18)$\mathrm{H}(9)$ coupling and long-range heterocorrelations of $3-\mathrm{H}$ with $\mathrm{C}-2, \mathrm{C}-10$ and $\mathrm{C}-18,4-\mathrm{H}$ with $\mathrm{C}-18,18-\mathrm{H}$ with $\mathrm{C}-1, \mathrm{C}-2$ and $\mathrm{C}-19$, and NH with $\mathrm{C}-2$. The $\mathrm{C}_{8}$ isoprenic side-chain was suggested by the $[\mathrm{M}-69]^{+}$EIMS fragment ion and ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY spectra. The missing portion $C(5)-C(8)$ could then be constructed from the ${ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR data, in particular from the coupling of $7-\mathrm{H}$ with both $3 \times 20-\mathrm{H}$ and $8-\mathrm{H} \beta$, and of the latter with $9-H$. The two moieties $C(9)-C(1)-C(4)$ and $C(5)-C(8)$ could be joined as in the flat structure $(-)-1$ for joalin on the basis of long-range ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ heterocorrelations $H(4)-C(5)$ and $H(9)-C(1)$.

The $\alpha$ position for $9-H$ was revealed by only small couplings with adjacent protons; $9-\mathrm{H} \beta$ would have required a coupling of $c a$. 11 Hz with $8-\mathrm{H} \alpha$. Strong (differentially derived) NOE between $7-\mathrm{H}$ and $3-\mathrm{H}$ indicated that these two protons must be on the same face of the nine-membered ring, thus requiring that $9-\mathrm{H}$ is on the opposite face. Small coupling between 3-H and 4H , suggesting a dihedral angle of $c a .90^{\circ}$, allowed us to assign the $\alpha$ position to O-4. NOE between $4-\mathrm{H}$ and $3 \times 17-\mathrm{H}$, and a complete conformational analysis ${ }^{7}$ of the isoprenic side-chain, allowed us to establish the relative configuration and preferred conformation of joalin, as represented by $1 \mathbf{1 a}$. This is in line with all xenicanes so far reported, ${ }^{1}$ apart from the incorporation of nitrogen and chirality at C-9 which are unique features of joalin.

If the signs of the Cotton effects for joalin $\left(\Delta \varepsilon_{259}+5.8\right.$ and $\Delta \varepsilon_{217.5}-39.6, \mathrm{CDCl}_{3}$ ) are interpreted as for $\gamma$-butenolides, according to assumptions for related cases, ${ }^{8}$ joalin may be assigned the configuration $3 S, 4 R, 9 R, 10 R, 18 R$. This accords with all other xenicanes of algal origin so far reported. ${ }^{1}$ With the caveat about this extrapolation, based on chiroptical data for very limited models, ${ }^{8}(-)-1$ also represents the absolute configuration for joalin.

Apart from the unique feature of nitrogen incorporation, joalin differs notably from previously known xenicanes ${ }^{1}$ in showing high $J_{3,10}{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ coupling, 10.4 Hz . Typically, dictyolactone shows $J_{3,10} \approx 0,{ }^{9}$ which, in accordance with the structure in the crystal, ${ }^{9}$ indicates the conformation of the C-3 chain represented in 2 , with $17-\mathrm{Me}$ pointing toward the side of the lactone ring. For joalin, the large $J_{3,10}$ coupling suggests, in agreement with the results from molecular mechanics calculations, ${ }^{7}$ that $17-\mathrm{Me}$ points toward the opposite side of the lactam ring 1a.

The only precedent for an oxygen bridge between $\mathrm{C}-4$ and C -18 in xenicanes is given by dictyotalide, isolated from an Okinawan collection of Dictyota dichotoma. ${ }^{10}$ Bearing a lactone carbonyl at $\mathrm{C}-18$ and a free aldehyde group at $\mathrm{C}-19,{ }^{10}$
dictyotalide may be thought to originate from a xenicane bearing a 4-hydroxy and C-18, C-19 aldehyde groups, followed by oxidation at $\mathrm{C}-18$. A precursor of the same type, with $\mathrm{C}-19$ oxidized to the carboxy group, may be imagined on the route to joalin by way of C-19 amide formation, enzyme-driven amide nitrogen attack at the C-18 aldehyde and aminoketalization by $4-\mathrm{OH}$.

## Experimental

The brown seaweed Dictyota sp., likely Dictyota ciliolata Sonder (ex. Kützing) ${ }^{11}$ according to Prof. A. Meinesz and Mr. R. Lemée, who retain a voucher specimen, collected at low tide in September 1991 at the Pointe de Senti à Joal, south of Dakar on the Senegal coast, was immersed in MeOH and homogenized. Solvent evaporation, hexane extraction ( 1.2 g residue), and flash-chromatography with hexane/EtOAc gradient elution, was followed by evaporation of fractions 33-37 (40 $\mathrm{cm}^{3}$ each) and HPLC of the residue on Merck Lichrosorb Si60 ( $25 \times 1 \mathrm{~cm}$ ) with hexane/isopropyl alcohol $97: 3,5 \mathrm{~cm}^{3} \mathrm{~min}^{-1}$, to give joalin ( - )-1, $t_{\mathrm{r}} 10 \mathrm{~min}$, as colourless, mildly cytotoxic on L1210 cell lines, semicrystals $\mathbf{4 . 5 \mathrm { mg } , 0 . 0 2 \% \text { based on dried }}$ seaweed residue); $[\alpha]_{\mathrm{D}}^{29}-59.0,[\alpha]_{434}^{20}-102\left(\mathrm{CHCl}_{3}, c 0.01\right) ; \dagger$ $\lambda_{\text {max }}(\mathrm{MeOH}) / \mathrm{nm} 205(12300)$ and $245 \mathrm{sh} ; \delta_{\mathrm{C}(\mathrm{TMS}=0 \mathrm{ppm})}\left(\mathrm{CDCl}_{3}\right.$, 75.43 MHz ) $162.60(\mathrm{~s}, \mathrm{C}-1), 131.91(\mathrm{~s}, \mathrm{C}-2), 49.72(\mathrm{~d}, \mathrm{C}-3), 94.60$ (d, J 154, C-4), 44.82 (t, C-5), 140.05 (s, C-6), 120.92 (d, C-7), 32.06 (t, C-8), 79.47 (d, J 148, C-9), 34.78 (d, C-10), 35.43 (t, C-11), 25.71 (t, C-12), 123.96 (d, C-13), 131.58 (s, C-14), 25.53 (q, C-15), 17.69 (q, C-16), 18.08 ( $\mathrm{q}, \mathrm{C}-17$ ), 85.70 (d, J 172, C-18), $174.60(\mathrm{~s}, \mathrm{C}-19)$ and $19.80(\mathrm{q}, \mathrm{C}-20) ; \delta_{\mathrm{H}(\mathrm{TMS}=0 \mathrm{ppm})}\left(\mathrm{CDCl}_{3}, 299.94\right.$ $\mathrm{MHz}, J$ in Hz) $2.77\left(\mathrm{~d}, J_{3,10} 10.4,3-\mathrm{H}\right.$ ), 4.70 (dd, $J_{4,5 \alpha} 3.6, J_{4,5 \beta}$ $2.2,4-\mathrm{H}$ ), 2.41 (dd, $\left.J_{\mathrm{gem}} 13.4, J_{5 \alpha, 4} 3.6,5-\mathrm{H} \alpha\right), 2.10$ (dd, $J_{\text {gem }} 13.4$, $J_{5 \text { B.4 }} 2.2,5-\mathrm{H} \beta$ ), 5.31 (br dd, $J_{7.8 \mathrm{z}} 12.7, J_{7,8 \mathrm{~B}} 3.9,7-\mathrm{H}$ ), 2.31 (ddd, $J_{g \mathrm{em}} 13.5, J_{8 \alpha, 7} 12.7, J_{8 \alpha, 9} 3.5,8-\mathrm{H} \alpha$ ), 2.56 (ddd, $J_{\mathrm{gem}} 13.5, J_{8 \beta} 7$ $\left.3.9, J_{8 \mathrm{\beta} .9} 3.1,8-\mathrm{H}_{\mathrm{B}}\right), 4.52\left(\mathrm{td}, J_{9,8 \mathrm{~B}} \approx J_{9,8 \alpha}=3.1, J_{9,18} 1.3\right.$, $9-\mathrm{H}), 3.32(\mathrm{~s}, 9-\mathrm{MeO}), 1.75\left(\mathrm{~d}\right.$ sext, $J_{10,3} 10.4, J_{10.17} \bumpeq J_{10,11}$ $6.5,10-\mathrm{H}), 1.15(\mathrm{~m}, 2 \times 11-\mathrm{H}), 1.85(\mathrm{~m}, 2 \times 12-\mathrm{H}), 5.00(\mathrm{t} \mathrm{sept}$, $\left.J_{13,12} 6.7,1.5,13-\mathrm{H}\right), 1.66(\mathrm{br} \mathrm{s}, 3 \times 15-\mathrm{H}), 1.57(\mathrm{br} \mathrm{s}, 3 \times 16-$ $\mathrm{H}), 1.01\left(\mathrm{~d}, J_{17.10} 6.5,3 \times 17-\mathrm{H}\right), 5.59(\mathrm{br} \mathrm{s}, 3 \times 18-\mathrm{H}), 1.58(\mathrm{~d}$, $\left.J_{20.7} 1.5,3 \times 20-\mathrm{H}\right)$ and $5.93(\mathrm{~s}, \mathrm{NH}) ; m / z$ (EIMS) (\%) $345(7$, $\mathbf{M}^{+}+$), $313\left(4[\mathrm{M}-\mathrm{MeOH}]^{+}\right), 277\left(28, \mathrm{M}^{+}-69-\mathrm{H}\right), 276$ (19 [M -69] ${ }^{+}$), 166 (50), 109 (39), 69 (66) and 41 (100); m/z
(HREIMS) $345.2303\left(\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{NO}_{3}\right.$ requires 345.2304), 313.2031 $\left(\mathrm{C}_{20} \mathrm{H}_{27} \quad \mathrm{NO}_{2}\right.$ requires 313.2042), 277.1672 ( $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{NO}_{3}$ requires 277.1678), 276.1602 ( $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{NO}_{3}$ requires 276.1600).

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$\dagger[\alpha]_{\mathrm{D}}$ Values recorded in units of $10^{-1} \mathrm{deg} \mathrm{cm}^{2} \mathrm{~g}^{-1}$.

